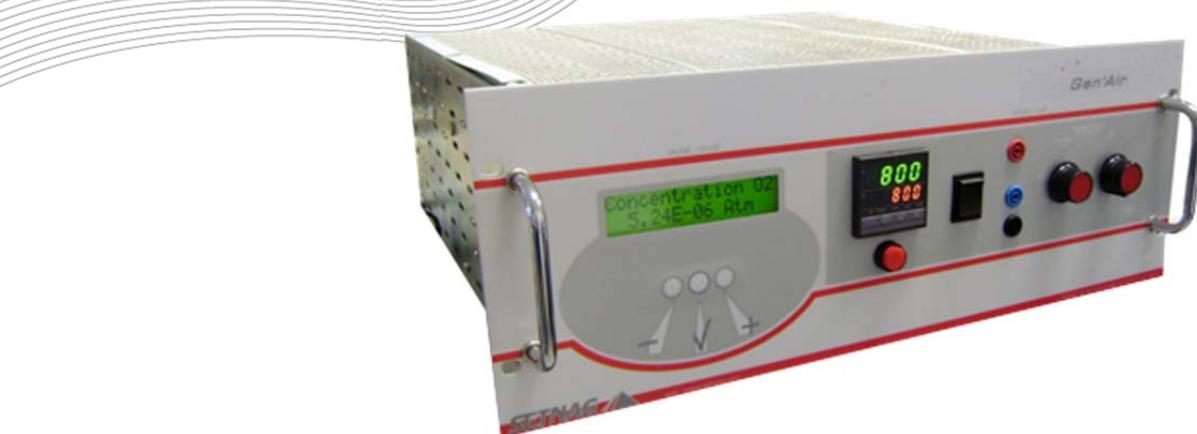


# GEN'AIR

## Oxygen pump-gauge



The GEN'AIR allows generating and measuring several different oxygen atmospheres. Its technology is based on the zirconia ionic conduction principle.

The GEN'AIR is made of two parts:

 **The pump:** it raises or decreases the oxygen partial pressure in the gas that passes inside its zirconia tube. It requires only a low gas flow: between 1 and 12l/h. It involves mixtures such as inert gas/oxygen or buffered mixtures/oxygen as CO/CO<sub>2</sub>/O<sub>2</sub> or H<sub>2</sub>/H<sub>2</sub>O/O<sub>2</sub>.

 **The gauge:** it measures the partial pressure generated by the pump. Thanks to the MicroPoas<sup>1</sup> its response time is very fast and it gives extremely accurate measurements.

<sup>1</sup> Patented design (University of Grenoble – France)

- **Generation and analysis of atmospheres at controlled oxygen rates**
- **Use of only small quantity of carrier gas**
- **Limited costs owing to the use of a single gas**
- **Large dynamic scale**
- **Compact and secured system**
- **Almost maintenance-free and low servicing requirements**
- **Extremely high measurement stability**

## Operation principle

### The pump:

A selector and a potentiometer are on the front panel to adjust the voltage applied to the pump, between 0 and around +/-1250mV. This generates an oxygen flow through the zirconia tube. The flow follows the Faraday's law:

$$X=X_0\pm 0,209\cdot I/D$$

Where  $X_0$  is the mole fraction of oxygen before the pump,  $X$  is the mole fraction of oxygen after the pump  
 $I$  is the current intensity in amperes,  $D$  is the flow of the carrier gas in l/h

### The Gauge:

The gauge is placed after the pump; it enables validating the partial pressure generated by the pump. The MicroPoas - zirconia sensor with built-in metal reference – carries out the measurement.

The MicroPoas is based on the Nernst's law, like all other zirconia:

$$E=(RT/4F)\ln(P_{mes}/P_{ref})$$

As for the MicroPoas, the reference partial pressure is set by an equilibrium between a metal and its oxide.

## Example of performances

At 1.6l/h and 800°C for a gas containing 5% oxygen in nitrogen:

Voltage applied to the pump in mV	Oxygen partial pressure in atm
200	3.70E-02
400	2.30E-02
625	5.40E-03
900	1.10E-08
-1265	1.40E-01

## Technical features

<b>Measurement range</b>	10 <sup>-35</sup> to 0.25atm*
<b>Necessary flow</b>	1 to 12 l/h**
<b>Output signals</b>	0-20mA or 4-20mA, linear, with galvanic insulation RS232 port
<b>Dimensions</b>	430x170x430 mm (wxhxd)
<b>Weight</b>	15 kg
<b>Power supply</b>	115 or 230 Vac – 50/60 Hz
<b>Power</b>	550 VA

\*\* Measurement of trace oxygen with a zirconia sensor remains delicate insofar as the presence of trace of combustible component impurities may create instability. More specifically inside the 10<sup>-6</sup> to 10<sup>-12</sup> atm O2 interval. The use of buffered mixtures enables generating reducing atmospheres under control.

\*\* The flow is controlled by an external system. We advise the use of a mass flow controller (please contact us).

*Specifications are subject to change – for improvement purposes – without notice.*

Technopole de Château-Gombert  
22-26, rue John Maynard Keynes - 13013 Marseille - FRANCE  
Tél : +33 (0)4 91 95 65 12 - Fax : +33 (0)4 91 64 22 27  
E-mail : contact@setnag.com

**setnag.com**

**SETNAG**   
EXPERT ET LEADER EN ANALYSE D'OXYGÈNE